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EcoTEK

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Ecological Truth & Environmental Research

Editor

Prof. Dr Snežana Šerbula

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PREFACE

The 31st international conference Ecological Truth & Environmental Research – EcoTER'24 focuses on showing the latest research findings and innovations in the field of ecology, environmental protection and sustainable development. The conference will be held in Sokobanja (Serbia) in hotel Sunce in the period of 18–21 June 2024.

The aim of the conference is to connect the experts in various fields in order to transform attitudes and behaviors in everyday practices, as well as in the industry and economy sector which is essential for achieving the desired changes that our society must undergo.

The 31st international conference Ecological Truth & Environmental Research – EcoTER'24 is organized by the University of Belgrade, Technical Faculty in Bor, and co-organized by the University of Banja Luka, Faculty of Technology; the University of Montenegro, Faculty of Metallurgy and Technology – Podgorica; the University of Zagreb, Faculty of Metallurgy – Sisak; the University of Pristina, Faculty of Technical Sciences – Kosovska Mitrovica and the Society of Young Researchers – Bor.

These Proceedings encompass 119 papers from the authors coming from the universities, research institutes and industries in 15 countries: Brazil, Norway, USA, Spain, Austria, Libya, Italy, Israel, Slovenia, Croatia, Romania, Bulgaria, Montenegro, Bosnia and Herzegovina, North Macedonia, and Serbia. It is a great honor and pleasure to cordially wish a warm welcome to all the participants of the conference.

As a part of this year's conference, the 6th Student Section – EcoTERS'24 will be held. We appreciate the contribution of the students and their mentors who have also participated in the conference and hope that students will continue to explore and to be curious, since education is a never-ending process, and knowledge is continuously growing.

The organization of the EcoTER'24 conference has been financially supported by the Ministry of Science, Technological Development and Innovation of the Republic of Serbia.

The support of the Donors and their willingness and ability to cooperate has been of great importance for the success of the EcoTER'24 conference. The organizing committee would like to extend their appreciation and gratitude to the Platinum donors of the conference – Serbia ZiJin Copper doo Bor and HBIS SERBIA, to the Gold donor of the conference – Elixir Group, as well as to the Silver donor of the conference – Serbian Chamber of Engineers.

We would like to express our sincere appreciation to all the authors who have contributed to the Proceedings. We would also like to express our gratitude to the members of the scientific, organizing and honorary committees, reviewers, speakers, chairpersons and all the conference participants for their support of the EcoTER'24. Sincere thanks go to all the people who have contributed to the successful organization of the EcoTER'24.

Prof. Snežana Šerbula,

President of the scientific and organizing committee



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STUDIES OF THE INFLUENCE OF GRAPHENE NANOSHEETS ON THE WETTABILITY OF ECO-FRIENDLY SOLDER ALLOYS

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Abstract

Lead-based solders are harmful to the environment and human health, so they must be replaced. Numerous studies have focused on the production of “eco-friendly” solder alloys. Recently, there has been an increasing number of researches dealing with lead-free composite solder alloys. This paper presents a review of research on the influence of graphene nanosheets (GNS) on the wettability of tin-based solder alloys. GNSs have been used to improve the physical and mechanical properties of solder. The variable mass content of GNS particles was successfully pressed into lead-free solder using a high-planetary ball mill. This method enabled better homogeneous mixing and consolidation of the mixed powder. After that, the powder that was obtained was sintered, and the effect of GNS on wettability was investigated. Increasing the mass content of GNS reduces the surface tension between the substrate and the composite solder, resulting in a lower contact angle and improved wettability.

Keywords: lead-free solders, graphene nanosheets, high-planetary ball mill, sintering.

INTRODUCTION

In the electrical industry, soldering is the most crucial connection method. The solders were heated to the melting point before soldering. The molten solder wet the substrate, creating connections between the liquid (solder) and the solid (substrate). The solder then solidifies as the joints cool, creating solder joints. Solders are often made of low melting point alloys since melting is a need for the soldering procedures. The most common solders are Pb-Sn alloys, and research into their characteristics is significant [1–4]. Lead is toxic and poses a danger to the environment and human health, therefore its elimination is necessary. Recognizing these facts, several nations have begun to take necessary safety measures, such as establishing rules limiting or prohibiting the use of lead in electronics. In this regard, the European Union (EU) adopted two directives: WEEE (Waste Electrical and Electronic Equipment) and RoHS (Restriction of the Use of Certain Hazardous Substances) [5,6].

There is no lead-free solder that can entirely outperform lead solder characteristics. It has been challenging to substitute lead-based solder, and research is ongoing. The new solder alloys must meet specific requirements in terms of melting temperature, wettability, mechanical, and electrical properties. Additionally, it must exhibit the right corrosion resistance in various conditions and at various temperatures. The lead-free solders that are used most frequently today are Sn-Cu, Sn-Ag, and Sn-Ag-Cu [7].

Reinforcements can be added to standard alloys to create a composite lead-free solder. There are many methods for reinforcing these solders, including powder metallurgy (PM),

melting and casting (MC), physical vapor deposition, precipitation, and others. PM includes mixing powders, compressing them at a certain pressure, and then sintering. The PM process is the most often used way of producing composite solder. Many researchers have chosen nano-sized particles for the reinforcements. As reinforcement, the following particles can be used: Si_3N_4 , Al_2O_3 , SiC , TiB_2 , TiO_2 , ZrO_2 , SnO_2 , and ZnO . Due to its remarkable properties, graphene has recently captured considerable attention from researchers. With superior mechanical properties, and electrical and thermal conductivity, graphene nanosheets (GNS) are expected to outperform carbon nanotubes (CNTs) and hold significant promise in the field of electronics [8]. To achieve better wettability and mechanical properties than the pure solder, Nai *et al.* [9] investigated the wettability of Sn-3.5Ag-0.5Cu solder reinforced with multi-wall carbon nanotubes (MWCNTs), produced by the PM method. The research results showed that with 0.07 mass.% MWCNTs in the matrix, the contact angle decreased from 29° to 24° , indicating an improvement in the wettability of the composite solder. The results corresponded with previous studies.

Solder wettability is a critical factor in determining the quality of the solder to substrate bond. Strong bonds between the solder and the substrate were promoted by good wetting. Wettability is the ability of a liquid to diffuse onto a solid surface in a specific environment, such as the atmosphere, temperature, and so on [10]. Wettability is defined by the contact angle (θ). Figure 1 illustrates the relationship between surface tension and contact angle, which can be described using the Young-Laplace equation (1):

$$\cos \theta = \frac{\gamma_{sa} - \gamma_{sl}}{\gamma_{la}} \quad (1)$$

The contact angle is the angle between the tangent drawn from point A to the circular line representing the contour of the solder drop and the surface of the substrate.

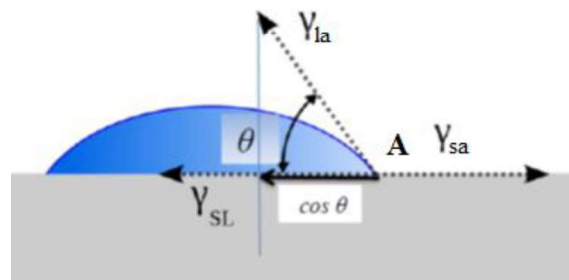


Figure 1 Schematic of contact angle (θ) of liquid metal on a solid surface [11,12]

When the contact angle (θ) is between 0° and 90° , the solder is considered to wet the substrate well. Better wettability is achieved with less contact angle, greater surface area, and spreading speed. Generally, a solder alloy with a smaller contact angle to the substrate provides a much more reliable interconnection during the soldering process [11,12].

This paper presents an overview of the research on the influence of GNS particles on the wettability of different tin-based composite solder alloys. The variable mass content of GNS

has been successfully integrated into lead-free solder using a high-planetary ball mill. This technique provided significantly greater homogenous mixing and consolidation of the mixed powder. Following that, the powder was sintered and the wettability was determined. The contact angle was measured using the sessile drop technique. The droplet was photographed and analyzed, and the contact angle between the solid surface, air, and the droplet was determined.

OVERVIEW OF RECENT RESEARCH

Yin *et al.* [13] used the contact angle method to measure the wettability of the Sn-0.3Ag-0.7Cu composite solder depending on the mass content of GNS. The result of the test shows that adding GNS increased solder wettability significantly. Figure 2 shows the influence of different mass content of GNS on the contact angle of the composite solder. The figure presents that the contact angle of the Sn-0.3Ag-0.7Cu solder decreased by approximately 29% with the addition of 0.01% GNS. With a further increase in the mass content of the reinforcement, the contact angle continued to decrease.

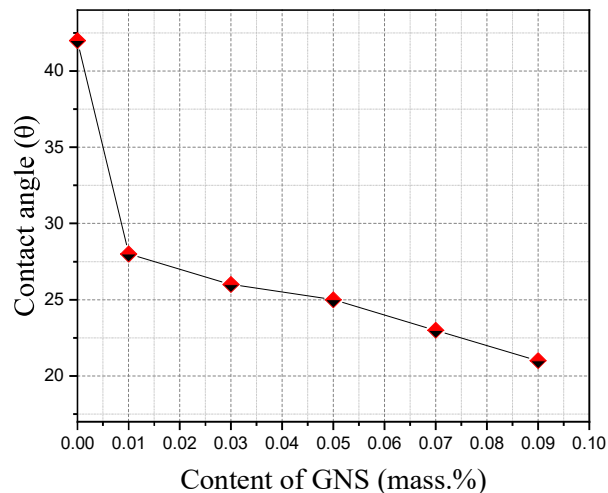


Figure 2 Contact angle of Sn-0.3Ag-0.7Cu-x GNS solder on the copper substrate [13]

After analyzing the results and reviewing previous research, they come to the following explanation. GNS are non-polar molecules with a hexagonal honeycomb lattice composed of carbon-carbon covalent bonds. When the GNS in the molten composite solder comes into contact with the non-polar organic acids in the rosin (solder paste) during soldering, the dispersion between the non-polar molecules of these substances can occur. The interphase tension between the composite solder and the rosin was reduced. When the interphase tension was reduced, the contact angle decreased, and the solder wettability was improved.

Liu *et al.* [14] studied the influence of GNS particles on the Sn-3Ag-0.5Cu composite solder and arrived at the following conclusions. As shown in Figure 3, the contact angle decreased as GNS content increased.

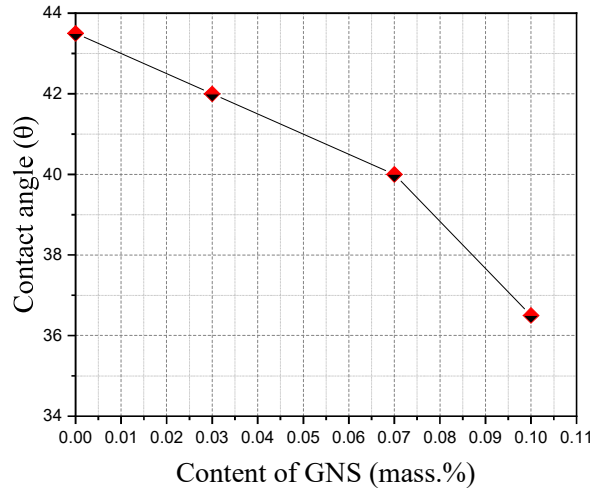


Figure 3 Contact angle of Sn-3Ag-0.5Cu-x GNS solder on copper substrate [14]

The composite solder with a mass content of GNS of 0.1% shows the smallest contact angle, which is 15.5% smaller than the starting solder. Analyzing the obtained results they came to the following conclusion. This improved wettability can be explained based on the effect of GNS particles on the interfacial tension between the composite solder and rosin during the soldering process. Through the soldering process, GNS should be dispersed at the solder/rosin interface. The presence of GNS can reduce the surface tension of the interface. Reduced interphase tension between the solder and the rosin causes a decrease in the contact angle. That improved the wettability of the composite solder.

Yin *et al.* [15] also studied the effect of different mass content of GNS on Sn-0.3Ag-0.7Cu composite solder and reached the following conclusions. In Figure 4, it can be seen that the contact angle decreases with the increasing mass content of GNS. The contact angle of Sn-0.3Ag-0.7Cu solder decreased by 31% by adding only 0.01% GNS particles. A further increase in the mass content of GNS up to 0.09% leads to a decrease in the contact angle. It follows that the wettability of the composite solder increases.

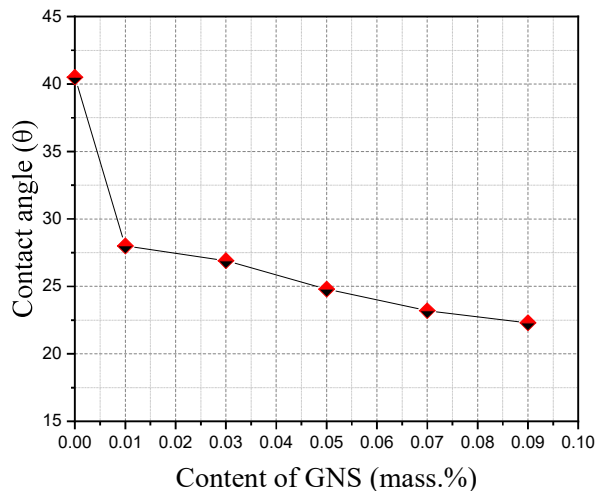


Figure 4 Contact angle of Sn-0.3Ag-0.7Cu-x GNS solder on the copper substrate [15]

According to the results, they reached the same conclusions as previous researchers. When the GNS particles present in the solder come into contact with the rosin, dispersion occurs between these compounds. The interfacial tension was reduced. Based on equation 1, reducing the interfacial tension between the composite solder and the rosin reduces the contact angle, which increases the wettability of the composite solder.

CONCLUSION

Solderability is the ability of solder alloys to wet and bond with a surface, forming a reliable and durable joint. Wettability tests were performed on several tin alloys reinforced with GNS particles, and the results were practically similar. The presence of GNS reduces the interfacial tension between the solder and the surface. As a result of the decreased contact angle, the wettability of the composite solder improved. By reviewing the literature and analyzing previous studies, it can be determined that the wettability of tin-based solder alloys reinforced with GNS increases.

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